



**ADVANCED SUBSIDIARY (AS)**  
**General Certificate of Education**  
**2022**

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**Biology**  
**Assessment Unit AS 3**  
*assessing*  
**Practical Skills in AS Biology**  
**[SBY31]**  
**FRIDAY 10 JUNE, AFTERNOON**

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**MARK  
SCHEME**

## **General Marking Instructions**

### ***Introduction***

The main purpose of the mark scheme is to ensure that examinations are marked accurately, consistently and fairly. The mark scheme provides examiners with an indication of the nature and range of candidates' responses likely to be worthy of credit. It also sets out the criteria which they should apply in allocating marks to candidates' responses.

### ***Assessment objectives***

Below are the assessment objectives for Biology.

Candidates should be able to demonstrate:

- AO1** Knowledge and understanding of scientific ideas, processes, techniques and procedures.
- AO2** Apply knowledge and understanding of scientific ideas, processes, techniques and procedures:
  - in a theoretical context
  - in a practical context
  - when handling qualitative data
  - when handling quantitative data.
- AO3** Analyse, interpret and evaluate scientific information, ideas and evidence, including in relation to issues, to:
  - make judgements and reach conclusions
  - develop and refine practical design and procedures.

### ***Quality of candidates' responses***

In marking the examination papers, examiners should be looking for a quality of response reflecting the level of maturity which may reasonably be expected of a 17 or 18-year-old which is the age at which the majority of candidates sit their GCE examinations.

### ***Flexibility in marking***

Mark schemes are not intended to be totally prescriptive. No mark scheme can cover all the responses which candidates may produce. In the event of unanticipated answers, examiners are expected to use their professional judgement to assess the validity of answers. If an answer is particularly problematic, then examiners should seek the guidance of the Supervising Examiner.

### ***Positive marking***

Examiners are encouraged to be positive in their marking, giving appropriate credit for what candidates know, understand and can do rather than penalising candidates for errors or omissions. Examiners should make use of the whole of the available mark range for any particular question and be prepared to award full marks for a response which is as good as might reasonably be expected of a 17 or 18-year-old GCE candidate.

### ***Awarding zero marks***

Marks should only be awarded for valid responses and no marks should be awarded for an answer which is completely incorrect or inappropriate.

### ***Marking Calculations***

In marking answers involving calculations, examiners should apply the 'own figure rule' so that candidates are not penalised more than once for a computational error. To avoid a candidate being penalised, marks can be awarded where correct conclusions or inferences are made from their incorrect calculations.

### ***COVID-19 Context***

Given the unprecedented circumstances presented by the COVID-19 public health crisis, senior examiners, under the instruction of CCEA awarding organisation, are required to train assistant examiners to apply the mark scheme in case of disrupted learning and lost teaching time. The interpretation and intended application of the mark scheme for this examination series will be communicated through the standardising meeting by the Chief or Principal Examiner and will be monitored through the supervision period. This paragraph will apply to examination series in 2021-2022 only.

/ denotes alternative points  
; denotes separate points

**Comments on mark values are given in bold**

*Comments on marking points are given in italics*

AVAILABLE  
MARKS

1	(a) A columnar epithelial; B goblet cell; C Paneth cell;	[3]	
	(b) Muscularis mucosa;	[1]	4
2	(a) (i) Respirometer;  (ii) To absorb CO <sub>2</sub> (so it doesn't affect the movement of the oil);  (iii) Must be airtight/equal volumes of KOH in each side;	[1] [1] [1]	
	(b) (i) Measure distance the oil moves in a set period of time; at 10°C and 25°C; practical detail, e.g. using a water bath or resetting oil levels by syringe;  (ii) Prediction – The rate of oxygen uptake will be greater at 25 °C than 10 °C; Explanation – Reactants have more kinetic energy; faster rate of respiration in the organisms;	[3] [3]	9
3	(a) Interrupted transect/or (random) sampling within each of the 4 dunes; method of random (dividing into a grid and random numbers) or systematic sampling within each dune; apparatus used in sampling, e.g. quadrat/pin frame/pin sampling or by description; further detail re-sampling, e.g. count numbers of <b>each</b> species/percentage cover/frequency; use of identification key;	[5]	
	(b) 66/200; 33%;	[2]	7
4	(a) Meiosis; 1 – homologous pairs/bivalent; 2 – chiasmata;	[3]	
	(b) Prophase 1;	[1]	4

		AVAILABLE MARKS
5	<p>(a) (i) As the concentration of sucrose solution increases the average % gain in mass decreases; [1]</p> <p>(ii) Water leaves the potato tissue by osmosis; as the <math>\psi</math> of the solution is more negative than the tissue; [2]</p> <p>(b) Potato <math>0.24 = -660 \text{ kPa}</math>; Carrot <math>0.53 = -1530 \text{ kPa}</math> (allow <math>-1520</math> to <math>-1540</math>); [2]</p> <p>(c) The carrot has a more negative/lower water potential than the potato; as the potato stores carbohydrate as starch which is insoluble; this doesn't affect the water potential of the potato cells; (accept converse for carrot) [3]</p> <p>(d) Any <b>pair</b> from:            Variable – Cut same size/diameter/length;            Explanation – SA will affect osmosis;</p> <p>Variable – method of drying samples before weighing;            Explanation – remove surface water/excessive drying prevent removing internal water;</p> <p>Variable – Temperature;            Explanation – will affect the rate of osmosis/KE of molecules involved;</p>	
		10
6	<p>(a) Rotate the eyepiece lens/move to higher magnification and it gets larger/eyepiece scale does not change size at higher magnifications/move stage micrometer/change focus slightly; [1]</p> <p>(b) <math>1 \text{ EPU} = 10 \mu\text{m}</math>; [1]</p> <p>(c) (i) Cell is <math>33 \text{ epu}</math>;  <math>33 \times 10 \mu\text{m}</math>  <math>= 330</math>; [2]</p> <p>(ii) Measure the length of several cells and take an average; [1]</p>	5

7	(a) (i)	<table border="1"> <thead> <tr> <th>Rate of reaction/arbitrary units</th></tr> </thead> <tbody> <tr><td>11.8</td></tr> <tr><td><b>25.6</b></td></tr> <tr><td>55.6</td></tr> <tr><td>15.6</td></tr> <tr><td>8.9</td></tr> </tbody> </table>	Rate of reaction/arbitrary units	11.8	<b>25.6</b>	55.6	15.6	8.9	one mark for correct values; one mark for correct number of decimal places	[2]	AVAILABLE MARKS						
Rate of reaction/arbitrary units																	
11.8																	
<b>25.6</b>																	
55.6																	
15.6																	
8.9																	
	(ii) Time for complete hydrolysis;			[1]													
(b)	Line graph, with points joined with straight lines; correct axis labels with units; points plotted correctly;			[3]													
(c) (i)	pH 7;			[1]													
(ii)	Compare – increasing acidity and alkalinity decreases the rate of reaction; Contrast – alkalinity has a more detrimental effect on enzyme rate of reaction;			[2]													
(d)	Line with peak lower than 55 at pH 7; but higher at extremes;			[2]	11												
The effect of pH on the rate of reaction of maltase																	
<p>The graph illustrates the relationship between pH and the rate of maltase reaction. The rate is highest at a neutral pH of 7, reaching a peak of approximately 56 units. At extreme pH levels (3 and 11), the rate is significantly lower, around 12 and 8 units respectively, demonstrating the enzyme's sensitivity to acidic and basic conditions.</p> <table border="1"> <thead> <tr> <th>pH</th> <th>Rate of reaction of maltase</th> </tr> </thead> <tbody> <tr><td>3</td><td>~12</td></tr> <tr><td>5</td><td>~26</td></tr> <tr><td>7</td><td>~56</td></tr> <tr><td>9</td><td>~16</td></tr> <tr><td>11</td><td>~8</td></tr> </tbody> </table>						pH	Rate of reaction of maltase	3	~12	5	~26	7	~56	9	~16	11	~8
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